

The Influence Mechanism of CSH-Montmorillonite Interface **Characteristics on Strength Properties of Cement-Stabilized** Montmorillonite

Ge Jinyu¹; Wang Qiang²; Peng Hu²; Zhou Juan²; Xu Fei¹; Qian Wenxun¹ 1.Nanjing Hydraulic Research Institute; 2. Nanjing Highway Development Center

Objectives

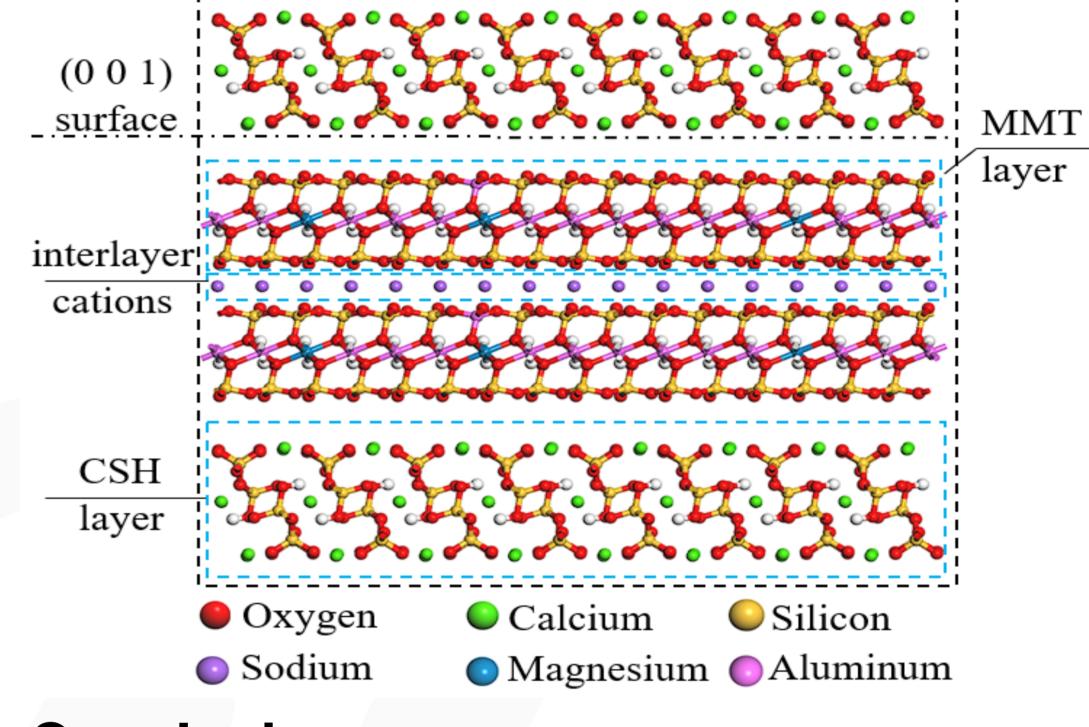
Cement-stabilized soil (cemented soil) technology is an important technical means to improve the poor soil properties such as mucky clay. The existing research on the development mechanism of cemented soil performance is mostly focused on the hydration of cement to characterize the products and microstructures. However, there is still a lack of analysis of the nanoscale interface bonding mechanism between hydration products and clay minerals in cemented soil and the interface energy source of cemented soil. Molecular Dynamics (MD) method can effectively explain the interactions and energy variations between different phases from the nanoscale. However, due to the need to consider the role of clay minerals, model construction and verification are more complicated. The current research focuses on the mechanical and structural evolution of a single material under different variables. There is little analysis of the dynamic mechanism of the interaction between cement hydration products and soil mineral interface areas in cemented soil, and the mechanism of performance improvement of cemented soil has not been fully elucidated.

Methods

- > The unconfined compressive strength (UCS) test was used to explore the effects of different water-to-cement ratio and cement-to-soil ratio on the macroscopic mechanical properties of cement-stabilized Montmorillonite.
- > The Molecular Dynamics (MD) method was used to quantitatively calculate the energy of the interface area between montmorillonite and CSH.
- > Grey relational analysis and X-ray diffraction (XRD) were used to explore correctness and deviation source of MD model.

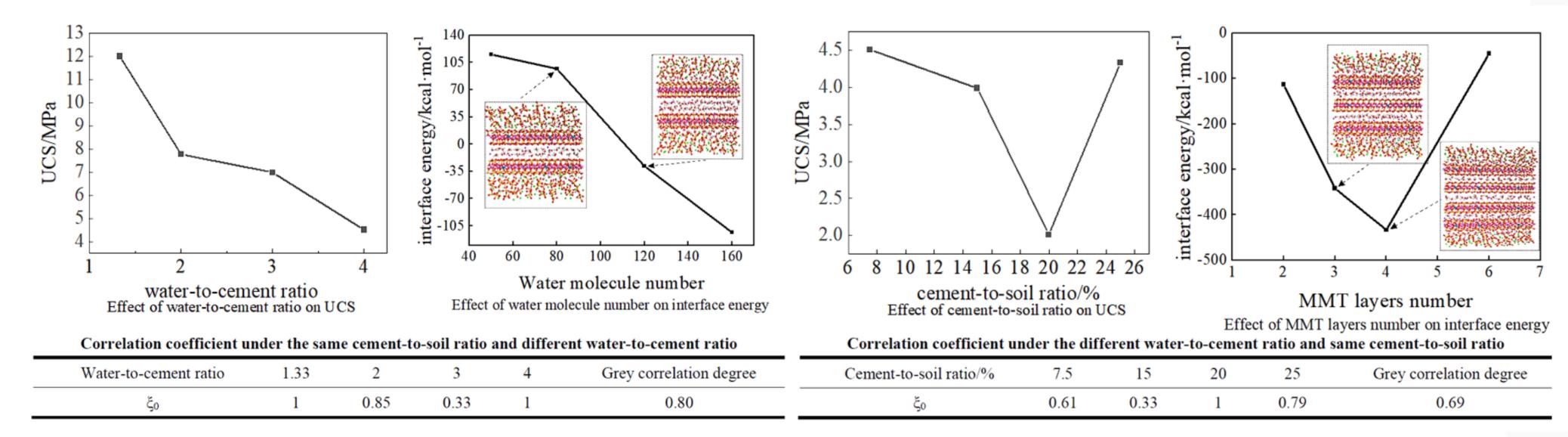
Results

Using grey relational analysis, we find that UCS has a good correlation with CSH-montmorillonite interface energy. The reason is that when the interface energy is high repulsive, the CSH-montmorillonite interface inhibits the dispersion of montmorillonite, resulting in higher stability of the system, which shows better mechanical properties, and vice versa.

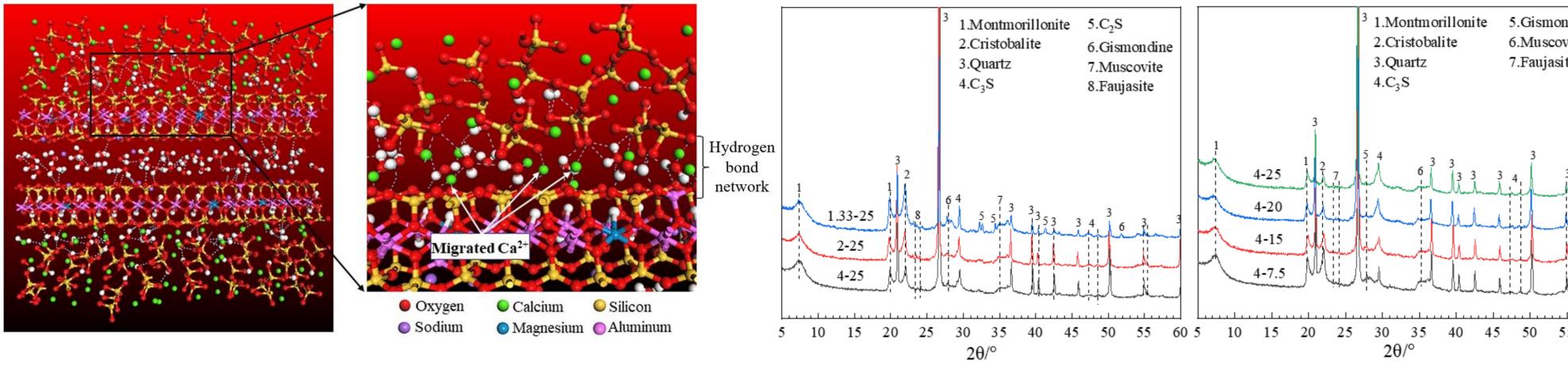


Conclusions

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Using the MD method, we find that the CSH-montmorillonite interface energy is mainly composed of van der Waals energy, hydrogen bond energy and electrostatic energy, and their lines and electrical properties are caused by the migration of cations.



		-		
³ 1.Montmorillonite	5.C ₂ S		³ 1.Montmorillonite	5.Gismondine
2.Cristobalite	6.Gismondine		2.Cristobalite	6.Muscovite
3.Quartz	7.Muscovite		3.Quartz	7.Faujasite
100			105	

- Ine CSH-montmorillonite Interface consists of van der Waals energy energy, hydrogen bond energy and electrostatic energy.
- > CSH-montmorillonite interface energy has good agreement with the UCS of the cemented soil, the interface energy decreases with increasing number of water molecules and decreases then increases with increasing number of layers of montmorillonite.
- ➤ The repulsive force at CSHthe montmorillonite interface be can maximized or the gravitational force minimized by adjusting the water-tocement ratio and the cement-to-soil controlling the strength ratio, thus performance of the cemented soil to

Using XRD test, we find that the migration characteristics of Ca²⁺ and Na⁺ are consistent with the results of MD simulation, which proves the correctness of MD model.

optimize it.

> XRD analysis results reflect that the migration characteristics of Ca²⁺ and Na⁺ under different water-to-cement ratio and cement-to-soil ratio are consistent with the MD simulation results.

